

***N*-ALKOXY-*N*-(DIMETHOXYPHOSPHORYL)UREAS AND *Z*-*N*-ALKOXY-1-(DIMETHOXYPHOSPHORYLOXY)BENZIMIDATES**

Shtamburg V.G.¹ *Dr. Sci., professor*, Kravchenko S.V.² *Ph.D., docent.*, Klots E.A.¹ *Ph.D., docent.*, Shtamburg V.V.¹ *Ph.D.*, Anishchenko A.A.³ *Ph.D., docent.*, Shishkina S.V.⁴ *Ph.D.*, Mazepa A.V.⁵ *Ph.D.*

¹*Ukrainian State University of Chemical Technology of Science and Technologies*

²*Dnipro State Agrarian and Economic University*

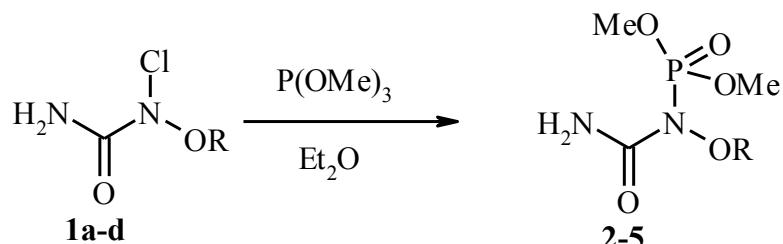
³*Oles Gonchar Dnipro National University*

⁴*Institute of Organic Chemistry of National Academy of Sciences of Ukraine*

⁵*A.V. Bogatsky Physico-Chemical Institute of National Academy of Sciences of Ukraine*

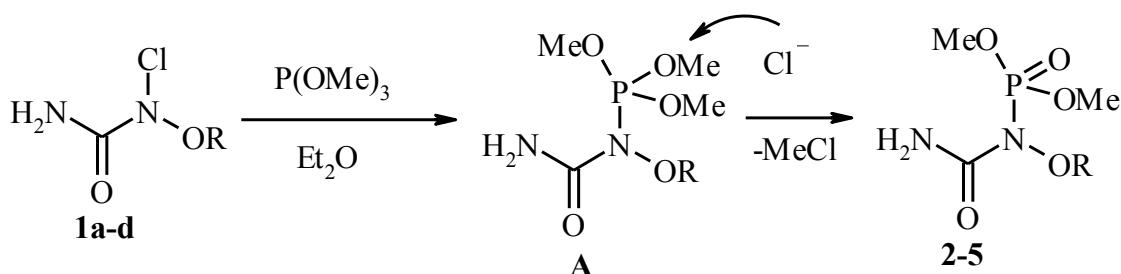
The possibility of the nucleophilic substitution of the chlorine atom in *N*-alkoxy-*N*-chloroureas and in *N*-alkoxy-*N*-chlorobenzamides allow to create the new reaction strategies that give access to such new biological relevant scaffolds

We have studied the interaction of trimethyl phosphite with *N*-alkoxy-*N*-chloroureas. The *N*-alkoxy-*N*-chloroureas **1a-d** react with trimethyl phosphite in ether selectively forming the *N*-alkoxy-*N*-(dimethoxyphosphoryl)ureas **2-5**.



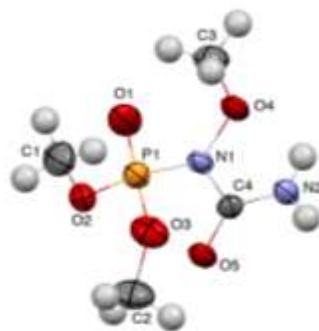
R=Me(**1a,2**), Et(**1b,3**), n-Bu(**1c,4**), i-Pr(**1d,5**)

It may be proposed this is another possible mechanism of *N*-alkoxy-*N*-phosphorylureas **2-5** formation.



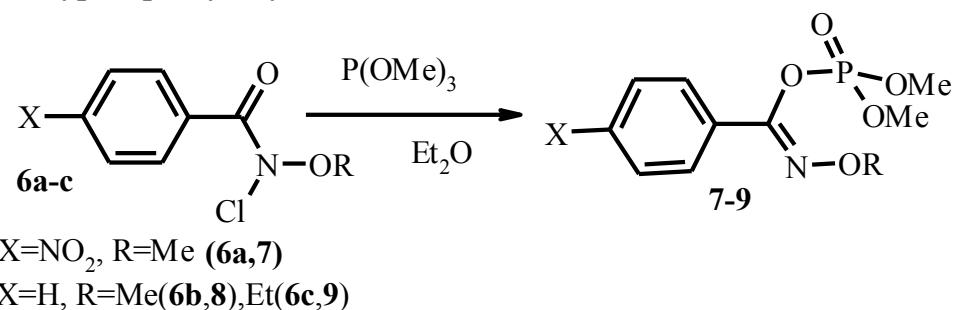
R=Me(**2**), Et(**3**), n-Bu(**4**), i-Pr (**5**)

The structure of *N*-alkoxy-*N*-phosphorylureas **2-5** has been proved by the ¹H, ¹³C, ³¹P NMR spectra and mass spectra. Also, the structure of compounds **2,4** has been confirmed by the XRD study.

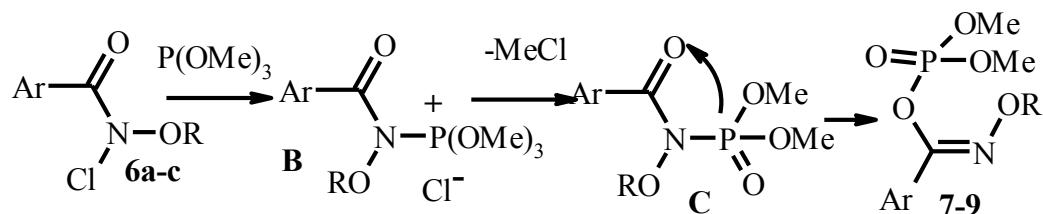


Molecular structure of *N*-methoxy-*N*-(dimethoxyphosphoryl)urea **2**.

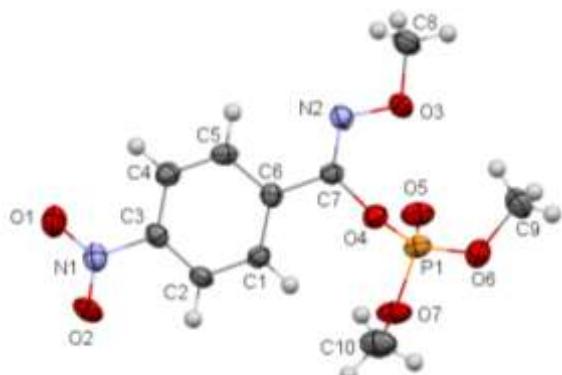
Also we have studied the interaction of trimethyl phosphite with *N*-alkoxy-*N*-chlorobenzamides. *N*-Alkoxy-*N*-chlorobenzamides **6a-c** selectively react with trimethyl phosphite in ether leading to the selective formation of *Z*-*N*-alkoxy-1-(dimethoxyphosphoryloxy)benzimidates **7-9**.



This is supposed to be another possible mechanism of compounds **7-9** formation. At the first stage the labile intermediates **B** formed. At the second stage the *O*-demethylation of the intermediates **B** by the chloride anion takes place (this is the new kind of Arbuzov reaction). It yields the unstable intermediates **C**. Then the N–O-migration of dimethoxyphosphoryl group takes place yielding compounds **7-9**. The driving force behind this migration could be the creation of a robust P–O bond.



The structure of *Z*-*N*-alkoxy-1-(dimethoxyphosphoryloxy)benzimidates **7-9** has been proved by the ^1H , ^{13}C , ^{31}P NMR spectra and mass spectra. Also, the structure of compound **7** has



been confirmed by XRD study.